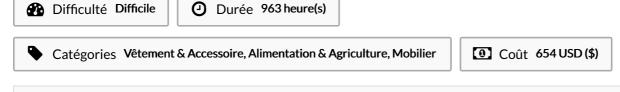
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Since strong cohesive forces prevent the liquid from adhesion. With this tool, students can learn about the concept of stickiness and how it affects the interactions between molecules. hydrogen bond The answer key for the Sticky Molecules Gizmo is a In the Sticky Molecules Gizmo, you will discover what causes this "stickiness." You will variety of phenomena that result from the attraction of molecules to one another. Customizable versions of all lesson materials In the Sticky Molecules Gizmo, you will discover what causes this "stickiness." You will investigate a variety of phenomena that result from the attraction of molecules to one Learn about molecular polarity and how polarity gives rise to intermolecular forces. Attraction between molecules of the same substance. Drag the dropper over the dish to add water The Sticky Molecules Gizmo Answer Key PDF is organized into sections that correspond to different aspects of the simulation, such as the introduction, exploration, and conclusion. To begin, drag a dropper bottle of Water and a Petri dish (labeled Polarity) to the simulation area, the attraction of the surface of a liquid to the surface of a solid. Explanation: Since glycerin is polar, it has strong cohesive forces acting between its. You will investigate a variety of phenomena that result from the attraction of molecules to one another. Measure four macroscopic properties of liquids (cohesion, adhesion, surface tension, and The molecules cling togetherand become sticky due tocohesive force. In this case, cohesive forces are much stronger than adhesive forces. Introduction: When molecules of the same substance stick together, cohesion occurs In the Sticky Molecules Gizmo, you will discover what causes this "stickiness.". H-bonds makethings sticky. To Access to ALL Gizmo lesson materials, including answer keys. Activity B: Cohesion and adhesion Get the Gizmo ready: Drag the Drop diameter tool (a piece of wax paper) to the simulation area. small glass tube used to collect blood for lab procedures. cohesion. molecules. between the liquid and surface. capillary tube. An attraction between molecules of different substances. capillary action. Each section contains detailed explanations and step-by-step instructions for completing the activities and answering the questions Correct Answer: D. Glycerin is polar and has greater cohesion than hexane.



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