

# Biochemistry of carbohydrates pdf

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Amines: condensation of alcohol (sugar) and ammonia. These are called amino-sugars. Chapter – Carbohydrates Chem Roper I. Carbohydrates – Overview A. Carbohydrates are a class of biomolecules which have a variety of functions: energy storage, Occurrence and importance. It is synthesized by plants through the process known as photosynthesis which converts more than billion metric tons of CO<sub>2</sub>. Carbohydrates are produced by the process of photosynthesis in which six carbon sugars or hexoses are produced using energy of sunlight, green pigment chlorophyll, CO<sub>2</sub> and H<sub>2</sub>O by Biochemistry of Carbohydrates Structure. Formerly the name carbohydrate was used in chemistry for any compound with the formula C<sub>m</sub>(H<sub>2</sub>O)<sub>n</sub>. CH Structure and Function Carbohydrates. Carbohydrates are commonly described as sugars, or saccharides, from the Greek word for sugar. Following this definition, some A carbohydrate is a biomolecule consisting of carbon (C), hydrogen (H) and oxygen (O) atoms, usually with a hydrogen-oxygen atom ratio of (as in water). B. Chemically speaking carbohydrates are polyhydroxyaldehydes, polyhydroxyketones, or compounds that yield them after hydrolysis. Named as sugar “amine” and when acetylated, the “N-acetyl” comes first Chapter – Carbohydrates Chem Roper I. Carbohydrates – Overview A. Carbohydrates are a class of biomolecules which have a variety of functions: energy storage, structure, other functions! Monosaccharides can be joined to make larger molecules. This is mainly because; the light energy from the sun is converted into Carbohydrates are the most abundant biomolecules on Earth. The simplest carbohydrates are called monosaccharides. An example is glucose. In biology, its ALWAYS at C. Often, the amino group will be further modified by an acetyl group (CH<sub>3</sub>CO-) making an amide. The carbohydrates comprise one of the major groups of naturally occurring biomolecules. Disaccharides contain two monosaccharides 7-It can be part of the body's extracellular ground substance (carbohydrate polymers lubricate skeletal joints) It can be associated with other biological macromolecules like proteins and lipids to form glycoproteins and glycolipids, respectively. It forms part of nucleic acids (ribose and deoxyribose in RNA and DNA, respect-) The carbohydrates are technically Monosaccharides: Derivatives.

 Difficulté Très facile

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## Sommaire

Étape 1 -  
Commentaires

Matériaux

Outils

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Étape 1 -

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