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Section 3.3 phase changes pdf answer key


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Melting, freezing, vaporization, condensation, sublimation and deposition are six common. Solid, liquid and gas. A phase change is a reversible physical change. Phase Changes. During a phase change, a substance rearranges the order of its particles (atoms or molecules). How are Phase Changes Melting, freezing, vaporization, condensation, sublimation, and deposition are six common phase changes. During a phase change the temperature of the substance does not change A phase change is reversible (can be undone or the substance can be returned to its original state) Phase change is a physical change, NOT A CHEMICAL CHANGE Phase Changes Temperature and Phase Changes Phase Changes that REQUIRE Energy (___)! Phase Changescontinued R_hsp09te_ChqxpFOCUS. ObjectivesDescribe phase changesExplain how temperature can be used to recognize a phase changeExplain what happens to the motion, arrangement, and average kinetic energy of water molecules during phase changesDescribe each of the six phase changes Section Phase ChangesPhase Changes. – Example: an iceberg floating in the ocean. How does the arrangement of water molecules change during melting and freezing? Vocabulary Reading Strategy. •Aphase change is the reversible physical Phase change is a ___ physical change that occurs when a substance changes from one state of matter to ____ Melting, freezing, vaporization, condensation, the phase change in which a substance changes from a gas to a solid without changing to a liquid first PHASE CHANGE: thermal energy (HEAT) is transferred to or from a substance that causes a change in phase. form of matter to another. Examples of phase change include melting (solid to liquid) and boiling (liquid to gas). phase changes. During a phase change the temperature of the substance does not phases of matter. Phase change is a ___ physical change that occurs when a substance changes from one state of matter to ___! Melting, freezing, vaporization, condensation, sublimation, and deposition are six common phase changes The most familiar example is water which melts at 0°C and boils at °C PHASE CHANGE: thermal energy (HEAT) is transferred to or from a substance that causes a change in phase. Key Concepts. that occurs when a substance goes from one.

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